THE STEREOCHEMISTRY OF FUKIIC ACID AND PISCIDIC ACID.

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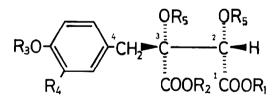
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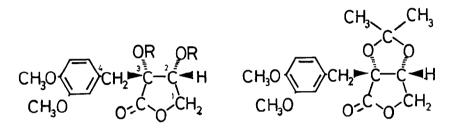
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Recently the structures of fukiic acid<sup>1)</sup> and piscidic acid<sup>2)</sup> have been established as 3,4-dihydroxybenzyl tartaric acid and p-hydroxybenzyl tartaric acid, respectively. In this communication we wish to report that these two analogous compounds have the same stereochemistry depicted as I and II.



 $\begin{array}{c} \text{I, } R_1 \sim R_3, R_5 = \text{H}; R_4 = \text{OH} \\ \text{OR5} \qquad \text{OR5} \qquad \text{II, } R_1 \sim R_3 = \text{H} \\ \text{III, } R_1 \sim R_3 = \text{CH}_3; R_4 = \text{OCH}_3; R_5 = \text{H} \\ \text{III, } R_1 = \text{CH}_3; R_2 \sim R_5 = \text{H} \\ \text{IX, } R_1, R_2 = \text{CH}_3; R_3 \sim R_5 = \text{H}; R_4 = \text{OH} \end{array}$ 

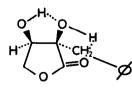
The relative configuration of fukiic acid was determined as follows. By means of  $N_{1,2}BH_4$ , fukiic acid dimethoxydimethylester (III), which was 3809 obtained by methylation of fukiic acid with diazomethane, was converted to a lactone (IV); ir  $\sqrt[V]{CCl_{4}}$  1782 cm<sup>-4</sup> (saturated  $\gamma$ -lactone); nmr § (d,acetone), 2.95 (2H, s, C<sub>4</sub>-H), 3.76 (6H, s, CH<sub>3</sub>O-), 4.0~4.3 (3H, m, C<sub>1</sub>-H and C<sub>2</sub>-H), 6.82 (2H, s, ArH), 6.92 (1H, s, ArH). The lactone IV was acetylated with acetic anhydride in pyridine to a diacetate V: m.p. 112~



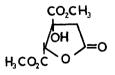
IV, R = HV, R = CH<sub>3</sub>CO

VI

113°:  $C_{47}H_{26}O_8$ ; M<sup>+</sup> m/e 352; ir  $\sqrt[KBr]_{max}$  1785 (saturated  $\sqrt[f]{-lactone}$ ), 1755 (acetyl c=0) cm<sup>-1</sup>; nmr § (CDCl<sub>3</sub>), 2.06 and 2.15 (3H, s, 2CH<sub>3</sub>CO-), 3.28 (2H, s, C<sub>4</sub>-H), 3.86 (6H, s, 2CH<sub>3</sub>O-), 3.85 (1H, d, C<sub>1</sub>-H, J=7.8 H<sub>Z</sub>), 4.08 (1H, d, C<sub>1</sub>-H, J=4.8 H<sub>Z</sub>), 5.48 (1H, q, C<sub>2</sub>-H, J=4.8 and 7.8 H<sub>Z</sub>), 6.82 (3H, s, ArH). Since the nmr spectrum of V has the signals at § 3.85, 4.08 and 5.48 due to ABX spin system, it is quite evident that the reduction was occured at carbomethoxy group attached at C-2 in compound TIT. The ir spectrum of the lactone IV under high dilute conditions ( $3.1 \times 10^{-4}$  mole)



A



VII

in carbon tetrachloride showed only one absorption band at 3575 cm<sup>-1</sup>, which indicates that absorption bands due to two kinds of hydrogen bonds arising from a moiety A are overlapped<sup>3)</sup>.

Further evidence for relative configuration was obtained from the formation of acetonide VI. m.p.  $135 \sim 136^\circ$ ,  $C_{16} H_{20} O_6$ . M<sup>+</sup> m/e 308;  $V_{max}^{\text{KBr}}$ 1778 cm<sup>-1</sup> ( $\gamma$  -lactone), which was prepared from IV on treatment with p-toluenesulfonic acid in acetone. These results suggest that the lactone IV has a cis diol and hence I has an <u>erythro</u> configuration.

In order to determine the absolute configuration of fukic acid, Horeau's method<sup>4)</sup> was employed. The lactone IV  $(1.87 \times 10^{-4} \text{ mole})$  was allowed to react with 2 -phenylbutyric anhydride  $(3.74 \times 10^{-4} \text{ mole})$  in pyridine. After hydrolysis, resultant 2 -phenybutyric acid showed a rotation of  $[d_{p}]_{p}^{z_{0}}$  +7.97 (c=3.97, benzene, optical yield 25.0%). Since the relative size of the substituent groups should be arranged in the order, H(S), -CH<sub>2</sub>-(M), HO- $c_{1}^{c}$ -CO<sub>2</sub>CH<sub>3</sub> (L), it is concluded that the stereochemistry at C-2 is S-configuration.

Further support of this result and the identity of absolute configurations between piscidic acid and fukiic acid were proved by exhaustive ozonolysis of these two compounds to hibiscus acid whose stereochemistry had been unequivocally established<sup>5</sup>. Thus ozonolysis<sup>6</sup> of piscidic acid monomethylester (VIII)(92.9 mg) and then treatment with methanolic hydrogen chloride yielded an oily substance (15 mg homogeneous on TLC); nmr  $\oint (d_b$ -acetone): 2.70 (1H, d, J=17 H<sub>X</sub>), 3.23 (1H, d, J=17 H<sub>Z</sub>), 3.73 (3H, s), 3.83 (3H, s), 5.32 (1H, s), 5.54 (1H, s); ORD  $\bigwedge \max_{max} 232 \text{ nm} \{\phi\} = \pm 10,800,$ CD  $\bigwedge \max_{max} 215 \text{ nm} \{\psi\} = \pm 12,500$ . These figures are compatible with those of hibiscus acid dimethylester (VII); (lit<sup>5</sup>., ORD  $\bigwedge \frac{MeOH}{max}$  230 nm ( $\psi$ ) = +10,887, CD  $\bigwedge \frac{MeOH}{max}$  214 nm ( $\vartheta$ ) = +14,525). The same method was applied to fukiic acid dimethylester (IX, 22 mg) to yield hibiscus acid dimethylester (VII) (3.1 mg), ORD  $\bigwedge \frac{MeOH}{max}$  232 nm ( $\psi$ ) = +11,100, CD  $\bigwedge \frac{MeOH}{max}$  215 nm ( $\vartheta$ ) = +13,900, exhibiting nmr spectrum identical with that of ozonolysis product from piscidic acid monomethylester.

All data above provide the evidence that both piscidic acid and fukiic acid have the same absolute configuration, S-configuration at C-2 and Rconfiguration at C-3.

## References and Footnotes

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- 3) If IV has a trans configuration, there should be appeared an absorption band due to free hydroxy group, cf. L.P. Kuhn. <u>J. Amer. Chem. Soc.</u>, <u>74</u>, 2492 (1952); E.L. Eliel and C. Pillar. <u>ibid.</u>, <u>77</u>, 3600 (1955).
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